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laboratory. 2. To learn the capabilities of the different types of balances which are available in the laboratory. 3. To relate the concept of significant numbers to the accuracy of mass and volume measurements. PRINCIPLES: One of the most important operations in a chemistry laboratory is the massing of

## **Applied Chemistry Chemistry 101 Laboratory Manual**

General Chemistry, Laboratory Manual CHM 1410L KOMAR UNIVERSITY OF SCIENCE AND TECHNOLOGY (KUST) 2 Preface and Acknowledgments The first version of General Chemistry I Lab Manual was drafted by Ms. Khalida Jaffar and Dr. Khalid Omer. It was used in the academic year, 2012-2013.

## **General Chemistry CHM 1410L**

General Chemistry II: Lab Manual:-utilizes Microsoft Excel® for data manipulation with a spreadsheet.-includes the PassPort®

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Troy University Chemistry Faculty. ... In this equation  $b$  is the light path length (1 cm for the equipment in this lab) and  $\epsilon$  is the molar absorptivity. This indicates that absorbance is proportional to concentration. ... General Chemistry II Lab Manual Last modified by:

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emeiidiliapl ns to Le Chatelier's Principle A highlighter upaup  
chisel Partner's Name: Part 1-Study of Concentration Effects: Oto  
Tolo  $\text{FeCl}_3 + 3\text{SCN}^- \rightarrow \text{Fe}(\text{SCN})_3 + 3\text{Cl}^-$  nocuom conmo sSin Test  
Effect on Color Tube Ion or Substance Added Effect on  
[ $\text{Fe}(\text{SCN})_3$ ] compared to Control Fea3 turned daixer red Innewed  
turned buue Fnciedsed black coro ...

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